



Structural and optical study of titanium dioxide thin films elaborated by APCVD for application in silicon solar cells

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Abstract. In this work, titanium dioxide (TiO₂) thin films have been deposited on glass and monocrystalline silicon by Atmospheric Pressure Chemical Vapor Deposition (APCVD) technique using titanium tetrachloride TiCl₄ as precursor. The structural, electrical and optical properties of the prepared TiO₂ thin films were evaluated by Atomic Force Microscopy (AFM), Four Point Probe (FPP) and Spectroscopic Ellipsometry (SE), respectively. These properties were exploited for application of the TiO₂ layers as antireflection coatings on monocrystalline silicon solar cells. Our experimental results show that the deposited TiO₂ thin films were polycrystalline, homogenous, compact and relatively smooth. The measured average optical transmittance of the TiO₂ films was about 85-90%. From the ellipsometry analysis, the refractive index of our TiO₂ thin films was found to be $n=2,25$ at the wavelength $\lambda=550$ nm, with a thickness of 56,2 nm. These experimental results obtained by APCVD are in excellent agreement with the computed results of the TiO₂ refractive index and thickness required for a high quality antireflection coating in industrial conditions. The obtained results demonstrate the real opportunity of the APCVD technique to prepare high quality antireflection coatings for crystalline solar cells. This indicates that the APCVD antireflection coatings may have a high potential industrial application.

Keywords: Titanium dioxide, APCVD, structural properties, optical properties, AFM, Ellipsometry, Four-Point Probe.

1. Introduction

Titanium dioxide (TiO₂) in thin films form has several properties that make it a material of interest for lots of applications. Indeed, TiO₂ films can be used as coating in anticorrosive protection, as catalysts in chemical industry and environmental purification phenomena. TiO₂ thin films have excellent electrical and optical properties, such as a high refractive index [1], [2] and excellent transmittance in the visible and near IR region of the solar spectrum [3], [4]. Since TiO₂ is highly transparent, it acts like a window for the transmission of solar radiations. Due to its interesting optical properties, these films have been widely used as antireflection coatings (ARC).

High quality antireflection coatings are required for the realization of high performance solar cells. One of the optical requirements for antireflection coating is high refractive index. This characteristic can be achieved by optimization of the film preparation technique. Several techniques have been developed for elaborating the TiO₂ thin films [5], [6], [7], [8]. Among them, atmospheric pressure chemical vapor deposition or APCVD is one of the most attractive, as it is known to be a low cost process giving high density stoichiometric and uniform films [9]. Moreover, it is especially

applicable to large area, continuous deposition, as required for growth on glass or crystalline silicon [9]. Due to its several advantages, CVD has become one of the main processing methods for the deposition of amorphous, monocrystalline and polycrystalline thin films and coatings for a wide range of applications.

In this work, TiO₂ thin films were fabricated on glass and monocrystalline silicon substrates by APCVD from the reaction of titanium tetrachloride TiCl₄ with oxygen. The structural, electrical and optical properties of the prepared TiO₂ thin films were evaluated by Atomic Force Microscopy (AFM), Four-point probe (FPP) and Spectroscopic ellipsometry (SE), respectively. In the light of these results, these properties were exploited for application of the TiO₂ layers as single-layer antireflection coatings (SLARC) on monocrystalline silicon solar cells.

According to [10], achieving low reflectance for monocrystalline silicon solar cells having TiO₂ as antireflection coating, is possible when the refractive index of the TiO₂ layer is equal to $n=2,2$ at $\lambda=550$ nm with an estimated thickness of 56,8 nm as illustrated in figure 1. These values can be achieved experimentally by optimization of the film preparation conditions.

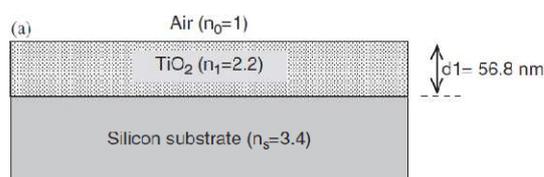


Figure1. Design diagram of a TiO₂ single-layer antireflection coating [10].

2. Experimental details

A. Preparation of TiO₂ thin films

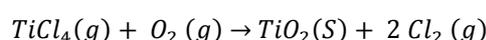
In this work, we have deposited the titanium dioxide thin films on glass and monocrystalline silicon using the APCVD technique. The titanium tetrachloride TiCl₄ (liquid) was used as the main precursor to form the titanium dioxide films.

The APCVD equipments consists of a Pyrex glass reactor with three inlets (for TiCl₄, O₂ and N₂), and a chemical vapor precursor supply system. The CVD reactor consists of a reaction chamber

equipped with a substrate holder, and a heating system with temperature control.

The glass and silicon substrates were washed successively in acetone, alcohol and distilled water to obtain clean surfaces which were dried before being placed in the reactor.

The TiCl₄ liquid precursor was heated in a tube connected to the reactor. The TiCl₄ vapor was transported into the reactor by oxygen as a carrier gas. The formation of the TiO₂ films is due to the reaction of TiCl₄ vapor with oxygen at the surface of the sample as follows:



A nitrogen gas (N₂) was injected into the chamber during the reaction in order to avoid undesirable oxidation of silicon at high temperature and to obtain a good uniformity of the films.

The CVD system being an “open” reactor, the resulting gases after deposition (Cl₂ in this case) were removed from the reactor.

A range of films were then grown at the deposition temperature of 450°C during 15 min and the oxygen pressure of 1 bar. Film thickness was controlled directly by deposition time. The deposited TiO₂ films on silicon and glass substrates at 450°C were then annealed at 450°C for a period of 1 hour.

B. Characterization methods

The surface morphology of the films was studied by Atomic Force Microscopy using the AFM apparatus of NT-MDT Solver (France) operating in the air at ambient conditions. Samples topography was imaged by scanning in the non-contact (NC-AFM) mode at resonant frequency of 310,75 kHz and a scan rate of 1 Hz. For comparison, the AFM measurements were also performed on bare silicon served as substrate and its RMS surface roughness was given. Roughness analysis and grain size measurements were performed using the Nanoscope software: “Nova”.

The optical properties of the deposited TiO₂ thin films were studied on the basis of the spectrophotometric measurements of the transmittance using the UV-VIS-IR

“spectrophotometer 0306029S1” in the wavelength range of 300-800 nm.

The refractive index of the prepared TiO₂ films was evaluated by ellipsometry measurements using the “WINELLI_II 2.2.0.6” ellipsometer (“SOPRA” software). The thickness of the TiO₂ layers was measured by AFM and ellipsometry for comparison.

The four-point probe (FPP) “KEITHLEY 236/237 TUTORIAL” in the linear probes configuration was used to measure the electrical resistivity of the prepared TiO₂ films. Once the probes were in contact with the surface, a constant current was passed through the outer pins while the corresponding voltage drop was measured across the inside pins. In this work, currents of 1-2 mA were employed.

3. Results and discussion

A. Morphological studies

Figure 2, illustrates the two-dimensional (2D) AFM topographic images of a Si substrate and annealed thin film of TiO₂ on Si. The AFM images reveal the polycrystalline structure and the homogenous surface of the prepared TiO₂ films. These films are dense, uniform and compact. They showed an excellent adhesion and passed the scotch Tape test with no visible degradation.

As estimated by AFM, the thickness of the TiO₂ film on silicon substrate was found to be 56,2 nm which is confirmed by ellipsometry measurements.

The surface roughness can be quantitatively identified by the root-mean-squared roughness (rms. Roughness: R_{rms}). R_{rms} is given by the standard deviation (S.D) of the data from AFM images, and determined using the standard definition as follows [11]:

$$R_{rms} = \sqrt{\frac{\sum_{n=1}^N (Z_n - \bar{Z})^2}{N-1}} \quad (1)$$

Where Z_n represents the height of the nth data, \bar{Z} is equal to the mean height of Z_n in AFM topography, and N is the number of data.

From our analysis, we observed a regular roughness and uniform grain sizes which varied from 30 nm to

34 nm for the as-deposited TiO₂ films and after annealing at 450°C for 1 hr, respectively. These values are in good agreement with the mean crystallite size determined as a function of deposition temperature in TiO₂ films, in other studies [12], [13].

In our work, the RMS surface roughness is determined from the area of 1,5 μm × 1,5 μm for all samples.

As it can be seen in figure 2, the bare silicon is relatively smooth and its characteristic RMS roughness is 2,5 nm. An RMS roughness of 5,1 was estimated for the as-deposited TiO₂ films at 450°C. After annealing (at 450°C for 1 hr), we observed that the surface roughness increases to R_{rms} = 7,97 nm due to an increase in grain sizes. According to [11], our TiO₂ films exhibit values of RMS roughness similar to those obtained by B.S. Richards.

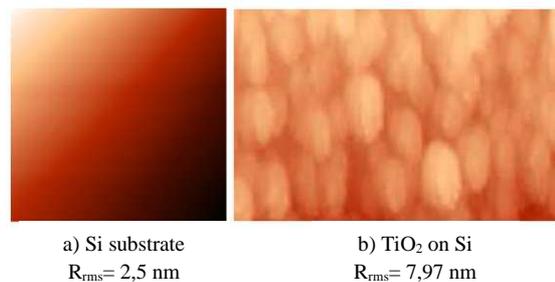


Figure 2: 2D-AFM images of:
a) Silicon substrate
b) TiO₂ thin films deposited by APCVD at 450°C after annealing at 450°C for 1 hr.

B. Optical studies

B.1. Optical transmittance

Information concerning optical transmittance is important for evaluating the optical performance of the TiO₂ thin films. The optical transmittance spectrum of the TiO₂ films deposited on glass at 450°C and annealed at 450°C for 1 hr is shown in figure 3.

As it can be seen, the TiO₂ films are optically transparent.

At the wavelength 350 nm, the transmittance of the deposited film was found to be 32%. A similar value was obtained by [14] for TiO₂ films using the APCVD technique.

The maximum transmittance obtained in this work for the TiO₂ films was estimated to 90% (at $\lambda = 800$ nm) which is in good agreement with the calculated transmittance into the TiO₂ films given by [15] as shown in table 1. The presence of the interference peaks in the transmission spectrum of TiO₂ films was also observed and clearly explained by [5].

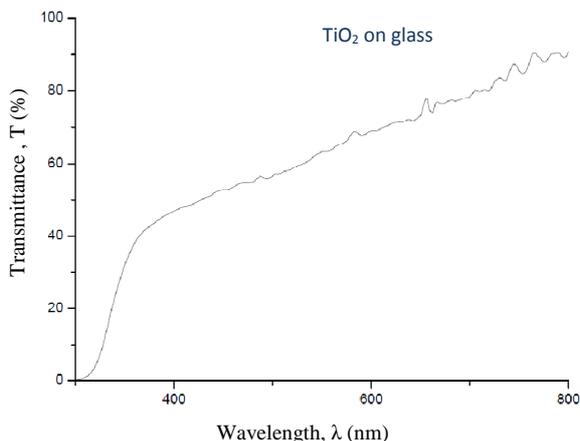


Figure 3. Optical transmittance spectrum of TiO₂ thin film on glass.

TABLE 1: Transmittance results comparison on TiO₂ thin films.

Wavelength λ	Transmittance T Experimental results obtained in this work	Transmittance T Simulation results given by [15]
800 nm	90,2 %	90%

B.2. Ellipsometry

The estimated variation of the refractive index versus wavelength for the deposited TiO₂ thin film on silicon substrate, and the corresponding best fit 2D graph are shown in figures 4 and 5 respectively. As mentioned above, the TiO₂ film thickness was 56,2 nm.

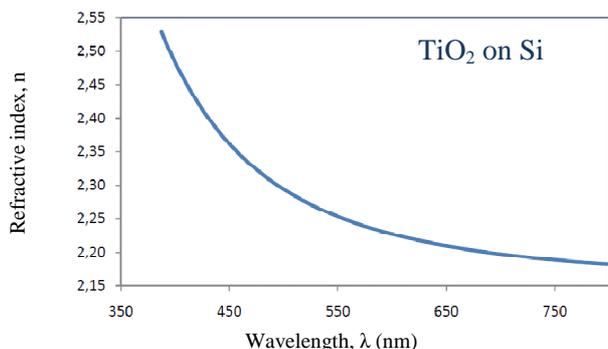


Figure 4. The variation of refractive index n versus the wavelength λ for TiO₂ Thin film deposited on Si at 450°C after annealing at 450°C for 1 hr.

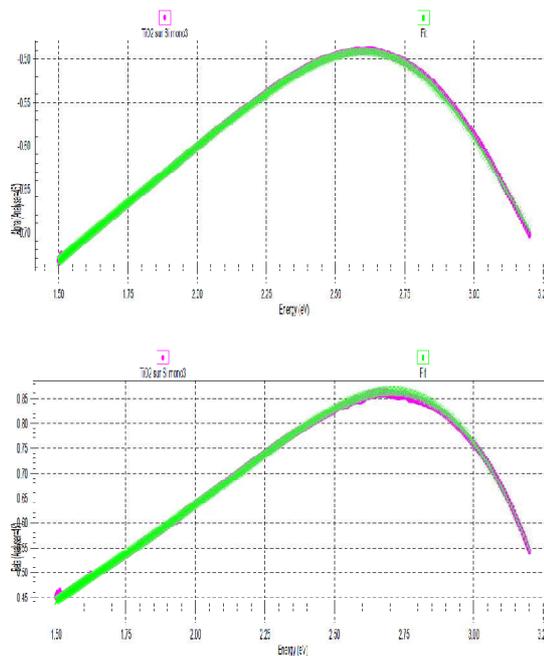


Figure 5. The corresponding best fit 2D graph

In this work, the refractive index of our TiO₂ film was found to be $n = 2,25$ at the wavelength $\lambda = 550$ nm which is in excellent agreement with the literature values for CVD TiO₂ films illustrated (by the dotted horizontal line) in figure 6.

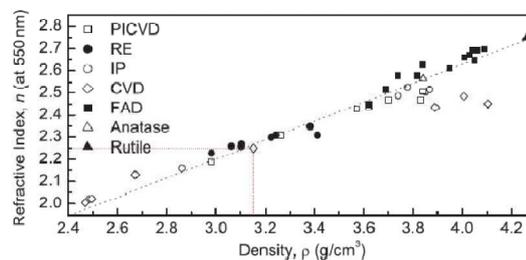


Figure 6. A linear relationship between values of TiO₂ film density and refractive index is observed for many films reported in the literature. (The dotted horizontal line shows the refractive index at 550 nm for the TiO₂ films elaborated by CVD [16]).

A decrease of the refractive index from $n = 2,52$ to $n = 2,18$ with increased wavelengths between 390 and 800 nm, was observed (figure 4).

To estimate the TiO₂ film density ρ in (g/cm³), the expression used is given below [16]:

$$\rho = \frac{n_f - 0,91933}{0,42751} \quad (2)$$

Additionally, the porosity φ (the volume of pores per volume of film) of the TiO₂ film depends on the refractivity of the film layer. It can be estimated by the equation (3) [16], [17], [18]:

$$\varphi = \left(1 - \frac{n_f^2 - 1}{n_d^2 - 1}\right) \times 100 \text{ (\%)} \quad (3)$$

Where n_f is the refractive index of the TiO₂ film, and n_d is the refractive index of pore free anatase TiO₂ which is found in the literature equal to 2,52 at 550 nm [17].

In this study, the density of the deposited TiO₂ film is calculated using equation (2) at 550 nm and is found to be $\rho = 3,11 \text{ g/cm}^3$.

The porosity of the prepared TiO₂ film which is assumed to be homogenous, is determined using equation (3) and found to be $\varphi = 24 \text{ \%}$.

TABLE 2: Ellipsometric results comparison.

Deposition technique	Deposition temperature	Refractive index obtained in this work ^a at 600 nm.	Refractive index given by [16] ^b at 600 nm.
APCVD	450°C	2,2	2,19

^aThe TiO₂ thin films are deposited from TiCl₄ at 450°C and then annealed at 450°C for 1 hour.

^bThe TiO₂ thin films are deposited from TPT $Ti(OC_3H_7)_4$ at 450°C and annealed at 1000°C for 1 hour.

As shown in table 2, the good agreement reached between our experimental values of refractive index and those obtained by B.S. Richards [16] in spite of the difference in the annealing temperature can be explained in terms of the growth rate of the films, which was faster in the case of TPT than the TiCl₄, as it was observed by Philip Evants [12].

C. Electrical properties

As mentioned above, the electrical resistivity of the deposited TiO₂ films was estimated with the Four-point probe, using the relation (5):

$$\rho = \frac{\pi}{Ln2} \left(\frac{\Delta V}{I}\right) e \quad (4)$$

Or

$$\rho = 4,53 \left(\frac{\Delta V}{I}\right) e \quad (5)$$

Where ΔV is the measured voltage, I the applied current and e is the thickness of the TiO₂ film which is equal to 56,2 nm.

The sheet resistance of the layer is then given by:

$$R_{\square} = 4,53 \frac{\Delta V}{I} \quad (6)$$

In this work, the electrical resistivity of the deposited TiO₂ films at 450°C annealed at 450°C for 1 hr, was found to be $\rho = 1,7 \times 10^{-3} \Omega \cdot \text{cm}$, which is in good agreement with the TiO₂ resistivity estimated by Ginley [19].

The sheet resistance of our TiO₂ films was equal to $R_{\square} = 303 \Omega/\square$.

It is interesting to note that the good conductivity of our TiO₂ films is very compatible with the conductivity requirements of TCO (Transparent Conducting Oxide) for application as ARC in silicon solar cells.

4. Application to antireflection coating

The above results show the possibility to fabricate TiO₂ layers with the optimal optical qualities required for antireflection coating, using the APCVD technique.

TABLE 3: Comparison between the experimental and calculated results of two parameters: refractive index and thickness of TiO₂ layers for application as ARC.

ARC parameters	Experimental results obtained in this work	Calculated results obtained by [10]
Refractive index n at $\lambda = 550 \text{ nm}$.	2,25	2,2
Thickness d	56,2 nm	56,8 nm

As shown in table 3, an excellent agreement is reached between our experimental results and calculated results given by Shui-Yang Lien [10] for TiO₂ single-layer antireflection coating (SLARC) on monocrystalline silicon solar cells.

5. Conclusion

In this work, we demonstrated by AFM studies that the TiO₂ thin films produced by APCVD technique from TiCl₄ as precursor at 450°C were polycrystalline, homogenous, compact and relatively smooth. The surface roughness of these films was observed to increase after annealing.

From the spectrophotometric measurements, the deposited TiO₂ thin films were optically transparent having an average optical transmittance of about 85-90%.

The ellipsometry analysis shows the good optical properties of our TiO₂ thin films having a refractive index of 2,25 at $\lambda = 550$ nm with corresponding thickness of 56,2 nm. These experimental results obtained by APCVD are in excellent agreement with the calculated results of the TiO₂ refractive index and thickness required for a high quality antireflection coating in industrial conditions.

Combining the high quality films deposition, lower cost processing, we believe that the APCVD elaborated TiO₂ antireflection coatings have good promise and potential to meet the goal of large-scale industrial applications.

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References

- [1] Lianchao Sun, Ping Hou, "Spectroscopic ellipsometry study on e-beam deposited titanium dioxide films", *Thin Solid Films* (2004), 525–529.
- [2] B.S. Richards, S.F. Rowlands, "Potential cost reduction of buried-contact solar cells through the use of titanium dioxide thin films", *Solar Energy* 76 (2004) 269–276.
- [3] P.S. Shinde, S.B. Sadale, "Properties of spray deposited titanium dioxide thin films and their application in photoelectrocatalysis", *Solar Energy Materials & Solar Cells* 92 (2008) 283–290.
- [4] R. Zallen, M.P. Moret, "The optical absorption edge of brookite TiO₂", *Solid State Communications* 137 (2006) 154–157.
- [5] P.S. Shinde, C.H. Bhosale, "Properties of chemical vapour deposited nanocrystalline TiO₂ thin films and their use in dye-sensitized solar cells", *J. Anal. Appl. Pyrolysis* 82 (2008) 83–88.
- [6] Masayuki Okuya, Koji Nakade, " Porous TiO₂ thin films synthesized by a spray pyrolysis deposition (SPD) technique and their application to dye-sensitized solar cells", *Solar Energy Materials & Solar Cells* 70 (2002) 425–435.
- [7] A. Bendavid, P.J. Martin, "Deposition and modification of titanium dioxide thin films by filtered arc deposition", *Thin Solid Films* 360 (2000) 241–249.
- [8] C. Legrand-Buscema, C. Malibert, "Elaboration and characterization of thin films of TiO₂ prepared by sol-gel process", *Thin Solid Films* 418 (2002) 79–84.
- [9] B. Vallejo, M. Gonzalez-Manas, "Characterization of TiO₂ deposited on textured silicon wafers by atmospheric pressure chemical vapour deposition", *Solar Energy Materials & Solar Cells* 86 (2005) 299–308.
- [10] Shui-Yang Lien, Dong-Sing Wu, "Tri-layer antireflection coatings (SiO₂/SiO₂-TiO₂/TiO₂) for silicon solar cells using a sol-gel technique", *Solar Energy Materials & Solar Cells* 90 (2006) 2710–2719.
- [11] B.S. Richards, "Novel uses of titanium dioxide for silicon solar cells", PhD thesis, University of New South Wales, Australia, 2002.
- [12] Philip Evans, Martyn E. Pemble, "Precursor-Directed Control of Crystalline Type in Atmospheric Pressure CVD Growth of TiO₂ on Stainless Steel", *Chem. Mater.* 18 (2006), 5750-5755.
- [13] Cullity, B. D. *Elements of X-Ray Diffraction*; Prentice Hall: Upper Saddle River, NJ 2001.
- [14] Yu Guo, Xi-wen Zhang, "Structure and properties of nitrogen-doped titanium dioxide thin films grown by atmospheric pressure chemical vapor deposition", *Thin Solid Films* 515 (2007) 7117–7121.
- [15] Yasuhiro Tachibana, Hitomi Y. Akiyama, "Optical simulation of transmittance into a nanocrystalline anatase TiO₂ film for solar cell applications", *Solar Energy Materials & Solar Cells* 91 (2007) 201–206.
- [16] B.S. Richards, "Single-material TiO₂ double-layer antireflection coatings", *Solar Energy Materials & Solar Cells* 79 (2003) 369–390.
- [17] N.R. Mathews, Erik R. Morales, "TiO₂ thin films – Influence of annealing temperature on structural, optical and photocatalytic properties", *Solar Energy* 83 (2009) 1499–1508.
- [18] R. Mechiakh, F. Meriche, " TiO₂ thin films prepared by sol-gel method for waveguiding applications: Correlation between the structural and optical properties", *Optical Materials* 30 (2007) 645–651.
- [19] D.S. Ginley and M.L. Knotek, "Hydrogen in titanium dioxide photoanodes", *Journal of the Electrochemical Society*, 126 (1979), 2163-2166.